

Class - B.Sc Part II (Honours)

Subject - chemistry

Topic - Name and Reaction

Paper - III C

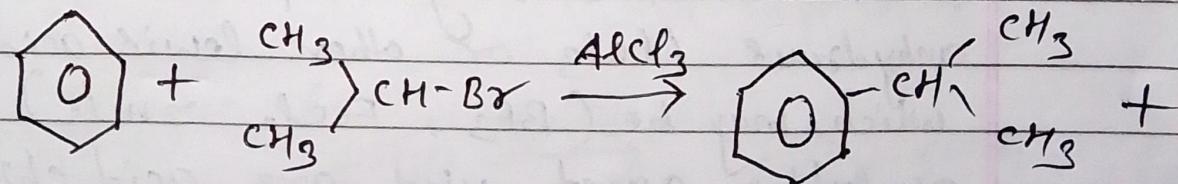
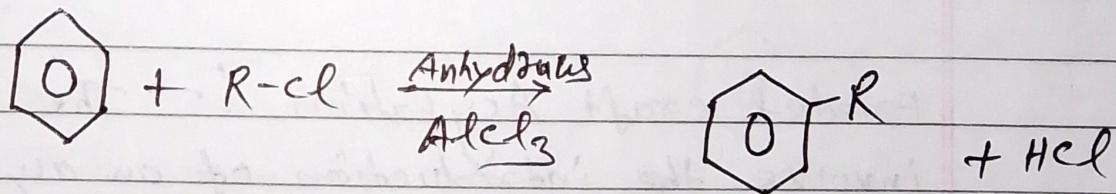
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H. D. Jain college, Agra

① Friedel-Crafts reaction! — The reaction involves the introduction of an alkyl group in Benzene for the synthesis of alkyl benzene.

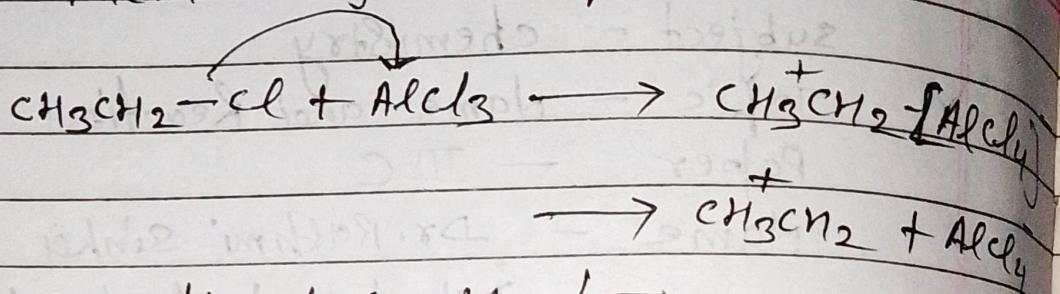
The alkylating agent used in Friedel-Crafts alkylation is alkyl halide the catalyst employed is Lewis acid which may be  $\text{AlCl}_3$ ,  $\text{BF}_3$ ,  $\text{FeCl}_3$



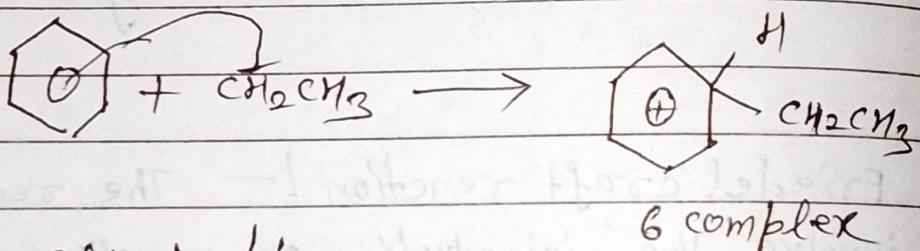
HBr

## Mechanism :-

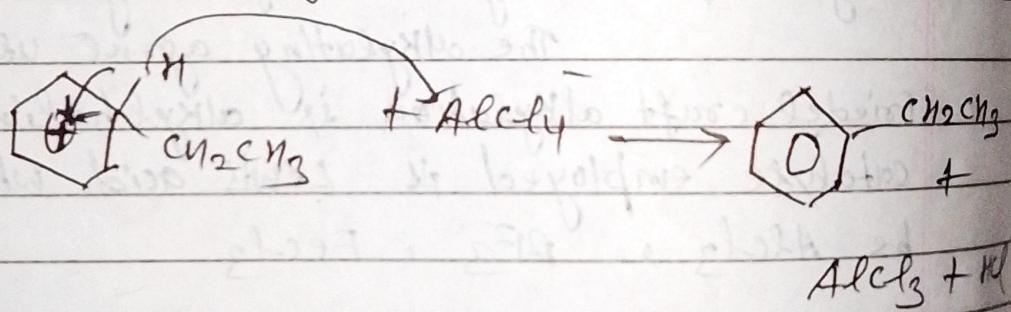
a. Generation of electrophile ! -



b. Formation of G complex ! -

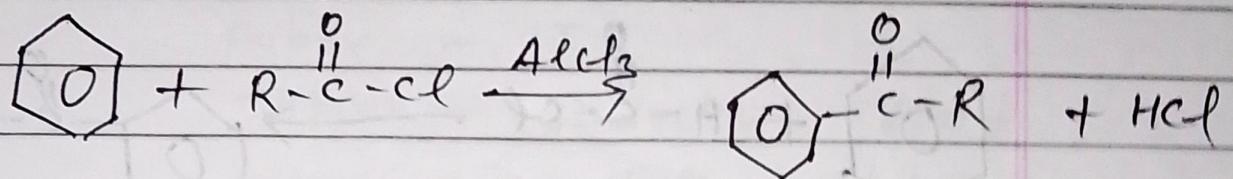


c. ~~Formation~~ elimination of Proton ! -

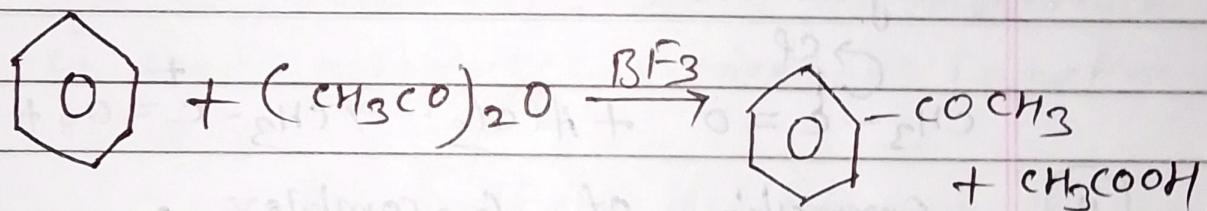
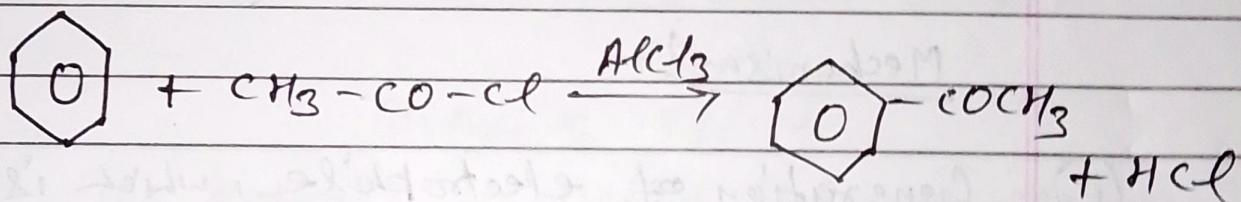


Friedel craft Acylation ! - The reaction involve the introduction of an acylgr (-RCO) in the aromatic ring in presence of anhydrous  $\text{AlCl}_3$  or other lewis acid catalyst which may be ( $\text{BF}_3$ ,  $\text{FeCl}_3$  etc.) the acylating agent used are acid chloride

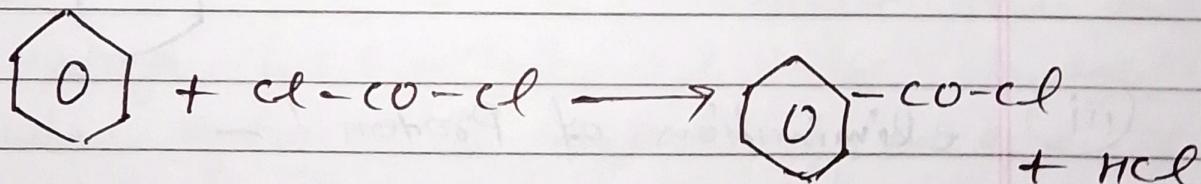
acid anhydride and ester.



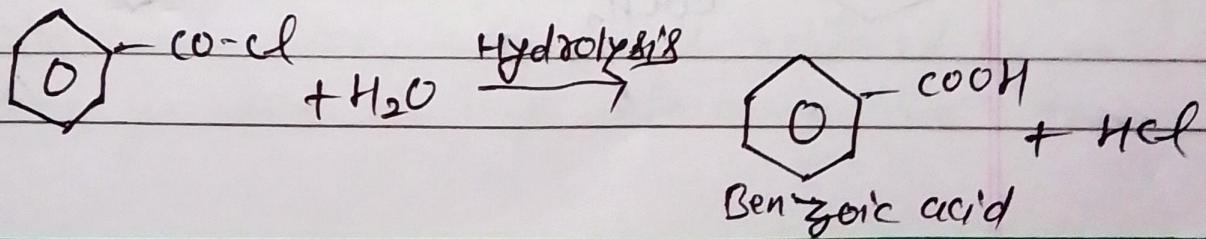
a. synthetic application of Friedel-Crafts acylation



b. synthesis of acid chloride and carboxylic acid / -

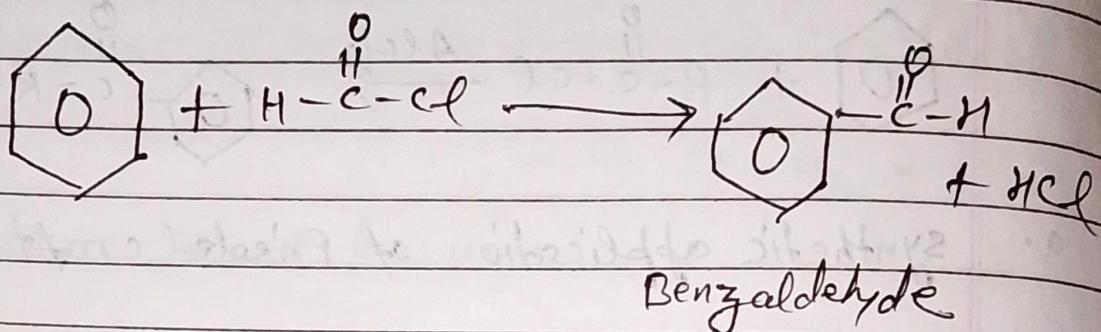


Benzoyl chloride  
(acid chloride)



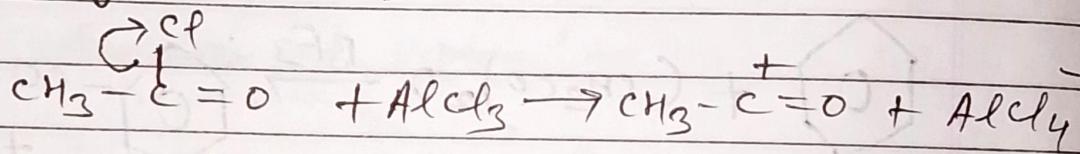
Benzonic acid

## C. synthesis of aldehyde (Formylation) / -

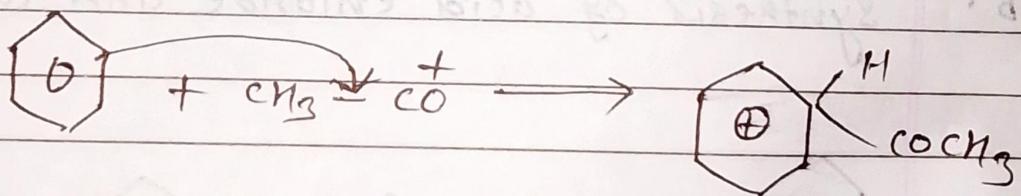


Mechanism! -

i) Generation of electrophile, which is an acylium ion! -



ii) formation of G-complex.



iii) elimination of Proton! -

